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ANSWER Answer the following questions in the space provided.

b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripld. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOìctbv-x

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If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B. gases are more compressible. C. the particles move faster in a gas and there is a greater distance between them.

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### Section 2 Answers

Chapter 11 Gases Review

Flashcards | Quizlet

Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

Chapter 11 - Gases

462 Chapter 11 Gases

Discovering the Relationships Between Properties If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider

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Changes in the properties of the gases (pressure, volume, temperature, or number of gas particles) inside and outside the balloon.

Chapter 11 Gases - An

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CHAPTER 11 REVIEW Gases

SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. b Pressure surf f a o c r e ce area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m<sup>2</sup> 3.

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this theory explains some of the  
properties of ideal gases. In this  
chapter, you will study the  
predictions of kinetic-molecular  
theory for gases in more detail.  
This includes the relationship  
among the temperature,  
pressure, volume, and amount of  
gas in a sample. SECTION 11.1  
Key Terms pressure newton  
barometer millimeters of mercury

SECTION 11.1 Gases and Pressure



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Pickford High School  
Section 2 Answers

CHAPTER 11 REVIEW Gases  
SECTION 2 SHORT ANSWER

Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase  
a. temperature increases, volume stays the same decrease  
b. volume increases, temperature stays the same

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SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. \_\_\_\_\_

Pressure = . For a constant force, when the surface area is tripled

the pressure is (a) doubled. (b) a

third as much. (c) tripled. (d)

unchanged. 2. \_\_\_\_\_ Rank the

following pressures in increasing order.

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