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set (32) Which one of the

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following reactions is NOT
balanced? $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$
 $2 \text{SO}_2 + \text{O}_2 \rightarrow 2 \text{SO}_3$ $\text{SF}_4 + 3$
 $\text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_3 + 4 \text{HF}$ $2 \text{KNO}_3 +$
 $10 \text{K} \rightarrow 5 \text{K}_2\text{O} + \text{N}_2$.

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ANSWER: Correct Chapter 9
Reading Question 2 Part A
Iron reacts with oxygen in
the air to produce rust
(Fe₂O₃) according to the
reaction shown below. 4 Fe +
3 O₂ → 2 Fe₂O₃ How many moles

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of rust can be formed if 1.8
mols of iron are present?

Hint 1. Begin with your
given number of moles.

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8 REVIEW Chemical Equations
and Reactions SECTION 3
SHORT ANSWER Answer the
following questions in the
space provided. 1. List four
metals that will not replace

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hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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2: Rank from most to fewest

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valence electrons of the following elements: Kr, Ga, Br, As, K. - Answer: Kr (8), Br (7), As (5), Ga (3), K (1). Valence electrons are equal to the atom's main group number. Link: <https://www.coursehero.com/file/2033>

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with flashcards, games and other study tools. On the periodic table, one of the columns within the main group elements; a family or group of elements exhibits similar chemical properties. (3.2). <https://quizlet.com/2>

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