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with Chapter 7, Section 7.2

1. How many valence electrons does a neutral magnesium (Mg) atom have?
2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so?
3. How many valence electrons does a single neutral chlorine (Cl) atom have?
- 4.

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uest 11 Name: Date: Hour: Information:
Energy Levels and Sublevels ... Considering your answer to question 5, how many electrons can fit in the entire 5th energy level? can (AveI is a ha AC5a) 50đ 7. Write the notation for an electron spinning clockwise in a p sublevel ...

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2. What is the relationship between the size of an s orbital and the principal energy level in which it is found?
3. What is the shape of a p orbital? How many p

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