

## Types Of Chemical Bonds Worksheet Answers Page 40

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A chemical bond is formed when electrons are shared between two atoms. There are three types of bonds: covalent bonds, polar covalent bonds and ionic bonds. The simplest example of bonding is demonstrated by the H<sub>2</sub> molecule. We can see that each hydrogen atom has a single electron from the periodic table.

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7.2 IONIC BONDING The chemical bond formed by transfer of electron from a metal to a non-metal is known as ionic or electrovalent bond . For example, when sodium metal and chlorine gas are brought into contact, they react violently and we obtain sodium chloride. This reaction is shown below:  $2\text{Na(s)} + \text{Cl}_2\text{(g)} \rightarrow 2\text{NaCl(s)}$

## 7 CHEMICAL BONDING

There are many types of chemical bonds that can form, however the 3 main types are: ionic, covalent, and metallic bonds. You must become familiar with how they work and the differences between the 3 types. I. Ionic bonding: Model 1 is a description of what chemists call ionic bonding. Ionic bonding occurs strictly between metal and nonmetal atoms.

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Accordingly, there are different types of chemical bonds, like, Ionic or electrovalent bond Covalent bond Co-ordinate covalent bond In addition to these we have a special kind of bond called hydrogen bond. Let us discuss about different types of bonds, their formation and the properties of the compounds so formed. 5.2 Ionic or Electrovalent Bond

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## *CHEMICAL BONDING*

There are two main types and some secondary types of chemical bonds: 1 Ionic bond Ionic bonding involves a transfer of an electron, so one atom gains an electron while one atom loses an electron. One of the resulting ions carries a negative charge (anion), and the other ion carries a positive charge (cation).

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